

# CHM 121IN Chemistry and Society REVISED Syllabus for Spring 2012

#### **Course Information:**

Course Prefix/Number: CHM 121IN

Semester: Spring 2012

Class Days/Times: TTh 11:30 a.m.-2:10 p.m.

Credit Hours: 4.0

Course Title: Chemistry and Society CRN (Section Code): 23462 Site/Room: SCI K207 Teaching Format: Integrated Lecture-Laboratory

#### Instructor Information:

Name:	David A. Katz
Office:	E-235 (Tortolita Building)
US Mail:	Pima Community College 2202 W. Anklam Rd. Tucson, AZ 85709-0270
Phone/Voice Mail:	(520) 206-6044
E-mail:	dkatz@Pima.edu
Web site:	http://www.chymist.com
Availability:	Office Hours: Office hours: MW 1:30-3:00 TTh 10:30-11:30; 2:30-3:00
	Generally, in addition to my office hours, I am in the office at least 30 minutes before or after class (if I am not in the lab). I am also available by appointment.

#### **Instructional Materials:**

**Required Text:** There is no required textbook for this course. You must have access to a computer with Internet connections. Course material or appropriate links to Internet sites will be available on the course web page at <u>http://www.chymist.com</u> under the heading **Pima Chem Courses** (left-hand menu) and then **Chem 121** (Left-hand menu). Some material will be supplied in class on a CD-ROM. The web files are in PDF format and require Adobe Acrobat Reader (available for free at *http://www.adobe.com*)

Laboratory Manual: Laboratory Experiments for Chem 121, Pima Community College. The experiments are available for download from the course web page at <u>http://www.chymist.com</u> - Pima Chem Courses (left-hand menu) - Chem 121 (Left-hand menu) The web files are in PDF format and require Adobe Acrobat Reader (available for free at http://www.adobe.com)

Laboratory Reference and Safety Manual: Katz, David A., The General Chemistry Laboratory Survival Manual, Hayden McNeil Publishing Co., 2006 (Textbooks are available at the West Campus bookstore. The PCC Bookstore can be accessed and books ordered via the Internet at www.Pima.bkstr.com.

#### Instructional Materials (continued)

**Explore the World of Earth Science** An Internet link to investigations and visualizations designed to accompany *Earth Science*, a high school textbook authored by Spaulding and Namowitz and published by McDougal Littell, is provided on the CHM 121IN web page. The Web site was developed by TERC, a non-profit educational research and development firm in collaboration with McDougal Littell.

**CD-ROM's** – Instructor Supplied in class at no charge Science and Science Fiction – Short stories and science articles Forensics – Hands-on forensics with additional readings

### Course Outline REVISED SCHEDULE

The following outline presents topics, activities, and experiments in the order they will be discussed in class. Although each topic is listed by specific date, there may be some changes due to extended class discussions or activities/experiments. **Applicable readings or links to reading material is available on the course web site** in addition to specific readings specified in the course outline, below. Science Fiction readings and Forensics readings and experiments will be supplied on CD-ROM. Problem assignments or calculations will be covered in context with activities and/or laboratory experiments.

Class topics are subject to change as issues become current in the news.

Laboratory experiments must be downloaded and read before class and the appropriate sections on laboratory techniques should be read before class. You will not be given copies of laboratory experiments in class.

Торіс	Lab Experiment and/or Activities
Welcome to class	No laboratory activities
Review syllabus and course requirements.	
Metric System (reading material on web site) Temperature (reading material on web site)	
Reading Assignment for next class: Explore the World of Earth Science (link on web site), Investigations page. Unit 7: Space Could Mars Support Life	
Are we alone? Elements and life in the universe	Build a Spectroscope
Life on Mars (and elsewhere in the universe) Origin of the elements Detecting elements in space	Lab Techniques: (Read for next class) <b>The General Chemistry Laboratory Survival Manual</b> Safety in the Academic Lab, Chapter 1 The MSDS, Chapter 2
Reading Assignment for next class: Element Symbols, Web site.	
Atoms and elements Element symbols The periodic table	Laboratory Orientation Check-in Lab safety
Reading Assignment for next class: Formula Writing (web site)	Lab Techniques: <b>The General Chemistry Laboratory Survival Manual</b> Laboratory Glassware and Apparatus, Chapter 4 Your Laboratory Drawer, Chapter 5
Quiz (in-class): Element names and symbols	Laboratory Safety Test
Chemical bonds	An Experiment in Alchemy: copper, silver, gold
	Lab Techniques: The General Chemistry Laboratory Survival Manual
Reading assignment for next class: Nomenclature of Organic Compounds (web site)	Laboratory Glassware: Its use, Care and Cleaning, Chapter 6 Volumetric Glassware: Graduated Cylinders, Burets, and Pipets, Chapter 7 The Bunsen Burner, Chapter 8 Mass Determination with Laboratory Balances,
Take-home quiz: Names and formulas of inorganic compounds	Chapter 9 Preparation of Synthetic Rubber (includes Preparation of a rubber ball from latex)
An introduction to organic compounds	
	Welcome to class         Review syllabus and course requirements.         Metric System (reading material on web site)         Temperature (reading material on web site)         Reading Assignment for next class:         Explore the World of Earth Science (link on web site), Investigations page. Unit 7: Space Could Mars Support Life         Are we alone? Elements and life in the universe and how they are detected         Life on Mars (and elsewhere in the universe)         Origin of the elements         Detecting elements in space         Reading Assignment for next class:         Element Symbols, Web site.         Atoms and elements         Element symbols         The periodic table         Reading Assignment for next class:         Formula Writing (web site)         Quiz (in-class): Element names and symbols         Chemical bonds         Inorganic compounds and nomenclature         Reading assignment for next class:         Nomenclature of Organic Compounds (web site)         Take-home quiz: Names and formulas of inorganic compounds

Date	Торіс	Lab Experiment and/or Activities
Feb. 2	Take-home quiz: Names and structures of organic compounds The nucleus and nuclear stability of atoms and isotopes	Nuclear chemistry Exp. 1. Determination of the Background Radiation Exp. 2. Determination of Half-life Exp. 3. Determination of the Type of Radiation Exp. 4. Inverse Square Law NOTE: Data will be shared between groups
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Feb. 7	Radioactivity, its applications and chemical and biological effects Radioactivity: Our Nuclear heritage Nuclear weapons Radiation we are exposed to every day (on web site)	Film: The Atomic Café Films to be viewed at home (links on web site): A Tale of Two Cities Hiroshima, Hirohito, & the Rising Sun Hiroshima-Nagasaki, August 1945
Feb. 9	The atmosphere Gases The greenhouse effect Reading assignment: Our atmosphere (web site) Our Changing Environment (web site) Explore the World of Earth Science (link on web site), Investigations page. Chapters 17-21 and Unit 5: Atmosphere and Weather	How Do We Affect the Quality of Our Atmosphere: Gases Simulating the Greenhouse Effect Cloud in a bottle Crush a can with air pressure Simulated sunset (Instructor Demonstration)
Feb. 14	Air pollution Acid rain: acids, bases, pH and the environment Reading assignments: Our Ozone Shield (web site) El Nino and climatic change (web site) Explore the World of Earth Science (link on web site), Investigations page. Unit 6: Can We Blame El Nino for Wild Weather?	How Do We Affect the Quality of Our Atmosphere: (continued) Acids, bases and pH Acid rain simulation (Instructor Demonstration) Effect of acid rain on our environment (Minerals and acid rain)
Feb 16	Ozone depletion and Chlorofluorocarbons What have we done to our environment?	Sunscreens: Preparation and Evaluation NOTE: Data will be shared between groups Lab Techniques: <b>The General Chemistry Laboratory Survival Manual</b> The Spectronic 20 UV-Visible Spectrophotometer, Chapter 13
Feb. 21	Water and water pollution Reading assignment: Explore the World of Earth Science (link on web site), Investigations page. Chapters 13-15, 22-24	Testing the Waters (Samples will be provided. You may bring in a 1 Liter [1 quart] sample of water from home to test.) NOTE: Data will be shared between groups Lab Techniques: <b>The General Chemistry Laboratory Survival Manual</b> The pH Meter, Chapter 14
Feb. 23	Rodeo Days – no class	
Feb. 28	The Earth: Soil structure and minerals Minerals and mining Reading Assignment: Explore the World of Earth Science (link on web site), Investigations page. Chapters 6, 12	Earth Science: Edible Soil Profile Mining for natural resources

Date	Торіс	Lab Experiment and/or Activities
Mar. 1	Metals and mining	Properties of iron
	Properties of Metals	Extracting a metal from its ore
Mar. 6	Soil structure (continued)	Soil Analysis
	Types of soil	
	Components of soil	
Mar. 8	. Energy sources and resources: Natural resources: coal, oil, and natural gas	Recycling a metal into a chemical compound: The preparation of Alum
	Conservation of resources: recycling, reuse	
	Nuclear energy and nuclear waste	Crystal growing (A homework assignment)
	Science Fiction CD will be given out today	
Mar. 12-18	Spring Break – no classes	
Mar. 20	Energy sources and resources: (continued)	Build a wind turbine
	Alternative sources: wind power, hydroelectric, geothermal, solar power	
	geothermal, solar power	
Mar. 22	Energy from chemical reactions:	Build a wind turbine (continued)
	Electrochemical cells and batteries	Complete construction and testing
Mar. 27	Instructor out of town – no class	
	Work on Report 1	
Mar. 29	Energy from chemical reactions (continued) Fuel cells	Batteries (Part 1)
	Solar energy	The Voltaic Pile The Electrochemical Cell
	Coldi onorgy	The Lemon Cell
Apr. 3	Nanotechnology	Batteries (Part 2)
	Viewing assignment:	The Storage Cell Construct a Dry Cell Battery
	What is Nanotechnology?	
	Report 1 is due today	Fuel Cells – build a mini fuel cell
Apr. 5	Nanotechnology	The Synthesis of Biodiesel from Vegetable Oil
	Discussion of articles/information	
Apr. 10	The science in science fiction Readings and discussions of the science	Magnetic Fluid (a nanotechnology project)
	behind the stories	Liquid crystals (a nanotechnology project)
Apr. 12	The science in science fiction (continued)	Construct a solar cell (A nanotechnology project)
	Readings and discussions of the science behind the stories	
	berning the stories	
Apr. 17	Forensic Science:	Faces: Identification of a suspect
	Crime Scene Investigation Material Evidence	
Apr. 19	Forensic Science:	Fingerprinting
	Fingerprinting	Fingerprints Dusting and lifting of fingerprints
		Cyanoacrylate development
Apr. 24	Forensic Science:	Ink Analysis
	Analysis of Inks and dyes Handwriting analysis	Handwriting Analysis
Apr. 26	Forensic Science:	Fibers and fiber identification: TIS fabric stain
	Fibers and fiber identification	Hair Analysis
	Hair identification	

Date	Торіс	Lab Experiment and/or Activities
May 1	Forensic Science Blood and blood drops	Blood drop studies
	The 2 <sup>nd</sup> report is due today	
May 3	Forensic Science: Drugs Poisons	Drug Testing of Some Over-The-Counter Drugs
May 8	Forensic Science DNA	DNA Isolation DNA Identification
May 10	Forensics Exam	Lab Check-out

## **Course Requirements:**

The final course grade will be based on attendance, class participation, quizzes, exams, lab experiments, projects, two short reports, and a final exam. The approximate percent weight of each is given below:

Attendance (includes participation in experiments)	10%
Short Reports (see schedule of important dates)	30%
Exams, quizzes, lab experiments, and projects	. 60%

# Spring 2012 Calendar of Important Dates

Jan. 17 Jan. 30	Spring classes begin Last day to withdraw with a refund
Feb. 21	Topic for 1 <sup>st</sup> Report due
Feb. 23-24	Rodeo Days – no classes
Mar. 1	45 <sup>th</sup> Day (Census day) Non-attending students will be withdrawn from the course
Mar. 27	Topic for 2 <sup>nd</sup> Report due
Mar. 12-18	Spring Break – no classes
Apr. 3	1 <sup>st</sup> Report due
Apr. 5	Withdrawal deadline
Apr. 19	Exam 3 – Write a science fiction story. Story is due no later than today.
May 1	2 <sup>nd</sup> Report due
May 8	Last day of regular classes. Last date to request a W grade in writing
May 9-15	Final exam week
May 10	Forensics Exam